Acta Crystallographica Section B Structural Science

ISSN 0108-7681

Ali Samy,^a‡ Robert E. Dinnebier,^a* Sander van Smaalen^b* and Martin Jansen^a

^aMax Planck Institute for Solid State Research, Heisenbergstrasse 1, D-70569 Stuttgart, Germany, and ^bLaboratory of Crystallography, University of Bayreuth, D-95440 Bayreuth, Germany

‡ Deceased.

Correspondence e-mail: r.dinnebier@fkf.mpg.de, smash@uni-bayreuth.de

Maximum entropy method and charge flipping, a powerful combination to visualize the true nature of structural disorder from *in situ* X-ray powder diffraction data

In a systematic approach, the ability of the Maximum Entropy Method (MEM) to reconstruct the most probable electron density of highly disordered crystal structures from X-ray powder diffraction data was evaluated. As a case study, the ambient temperature crystal structures of disordered α - $Rb_2[C_2O_4]$ and α - $Rb_2[CO_3]$ and ordered δ - $K_2[C_2O_4]$ were investigated in detail with the aim of revealing the 'true' nature of the apparent disorder. Different combinations of F (based on phased structure factors) and G constraints (based on structure-factor amplitudes) from different sources were applied in MEM calculations. In particular, a new combination of the MEM with the recently developed charge-flipping algorithm with histogram matching for powder diffraction data (pCF) was successfully introduced to avoid the inevitable bias of the phases of the structure-factor amplitudes by the Rietveld model. Completely ab initio electron-density distributions have been obtained with the MEM applied to a combination of structure-factor amplitudes from Le Bail fits with phases derived from pCF. All features of the crystal structures, in particular the disorder of the oxalate and carbonate anions, and the displacements of the cations, are clearly obtained. This approach bears the potential of a fast method of electron-density determination, even for highly disordered materials. All the MEM maps obtained in this work were compared with the MEM map derived from the best Rietveld refined model. In general, the phased observed structure factors obtained from Rietveld refinement (applying F and G constraints) were found to give the closest description of the experimental data and thus lead to the most accurate image of the actual disorder.

1. Introduction

Rietveld refinement is considered to recover the maximum amount of information that is contained in powder diffraction data (David et al., 2002). Nevertheless, this amount is smaller than that contained in a complete set of observed structurefactor amplitudes (van Smaalen, 2007). The correlations between the different crystallographic parameters as well as the application of constraints and restraints is often unavoidable in Rietveld refinement. In the case of an ordered crystal structure, the refinement process is able to reconstruct the phases of structure factors with reasonable accuracy even if the structure model is not perfect. However, if the crystal is disordered (except for simple cases) the refinement process is not easy, typically requiring the introduction of rigid bodies, constraints, restraints and anharmonic anisotropic displacement parameters (a.d.p.s) to reach convergence. In such cases the structural model obtained from Rietveld refinement needs to be verified and improved. This can be achieved by the

© 2010 International Union of Crystallography Printed in Singapore – all rights reserved Received 3 August 2009 Accepted 7 December 2009 Maximum Entropy Method (MEM), which can be used as a complementary method for deriving the most probable electron density from limited information by maximizing the entropy. Recent progress in synchrotron powder X-ray diffraction techniques, including third generation synchrotron sources, low-noise image-plate detectors (IP), new optical systems and analytical methods, enables the extraction of structure-factor amplitudes with high accuracy. These developments not only increase the success rate of crystal structure determination from powder diffraction data, but also (by determining the charge-density distribution using the MEM) allow for the investigation of advanced structural features such as disorder, diffusion pathways in ionic conductors, electron density due to chemical bonds and nano-applications (Dinnebier et al., 1999; Yashima & Tsunekawa, 2006; Takata, 2008).

However, the reconstruction of accurate electron density from experimental data suffers from model biasing effects in addition to the artifacts caused by the incompleteness of the data set. The dependence of MEM electron densities on the lack of completeness of the underlying data set was studied by Takata & Sakata (1996), and Yamamoto et al. (1996). Efforts have been undertaken to overcome these limitations by introducing alternative weighting factors that force the distribution of the residuals of the final structure factors towards the required Gaussian distribution (e.g. de Vries et al., 1994; Palatinus & van Smaalen, 2002). In particular for powder diffraction data, severe overlapping (where only the sum of the individual intensities of the overlapping peaks is available) can be handled using so-called G constraints (Sakata et al., 1990), thus avoiding a model bias due to the separation of the measured intensity into contributions from the reflections belonging to this overlap group. The combination of the MEM and Rietveld methods was introduced by Takata et al. (1995) and is called the REMEDY cycle, after the name of the computer program. In this method, the structure model is iteratively improved by replacing the values of the calculated structure factors (F_{calc}) from Rietveld refinement by the corresponding values of the observed structure factors $F_{\rm obs}$ obtained from MEM calculations.

Here we present a new combination of the MEM and the method of charge flipping (Oszlányi & Sütő, 2004, 2005, 2007; Palatinus, 2004). Reflection phases from charge flipping were introduced for two purposes: first to improve the accuracy of the phases obtained from the Rietveld method and second as a fast method to visualize the type of disorder independent of the Rietveld model. This approach not only succeeded in revealing the basic features of the crystal structure, but also fine details as the type of disorder (rotational and/or conformational disorder) and the type of thermal vibrations. In general, different types of structure-factor amplitudes and phases imposing different types of constraints were subjected to the MEM with the aim of acquiring the least biased electron density in order to visualize and understand the disorder of α - $Rb_2C_2O_4$ and α - Rb_2CO_3 (Dinnebier *et al.*, 2005). For comparison between the different MEM maps, the difference MEM maps $[\rho_{(I)}^{\text{MEM}} - \rho_{\text{calc}}^{\text{MEM}}]$, where (I) refers to F_{obs} , $F_{\text{obs}} + G$ constraints, $F_{\text{LeBail}} + G$ constraints, CF + $F_{\text{LeBail}} + G$ constraints and CF + $F_{\text{obs}} + G$ constraints, were calculated and visualized in three- and two-dimensional space, in addition to calculations of the agreement factors and the volume and area of the enclosed surfaces.

It will be shown that charge flipping with structure-factor amplitudes obtained from a Le Bail fit is a powerful combination to reconstruct an approximate electron-density map of disordered materials by the MEM. In addition, G constraints are used to handle the overlapping reflection in the MEM calculation, and the histogram matching method is used for repartitioning the overlapping reflections in the pCF approach.

2. Methods

2.1. Maximum Entropy Method (MEM)

The concept of entropy was introduced in the field of crystallography to handle the series termination effects in Fourier maps. A perfect Fourier map would require a complete set of structure factors up to at least $(\sin \theta/\lambda)_{max} = 5.0 \text{ Å}^{-1}$ (de Vries *et al.*, 1996). In the case of the powder diffraction experiment, accessible information is limited compared with single-crystal data owing to the projection of the three-dimensional reciprocal space onto the one-dimensional 2θ axis and the resulting intrinsic and accidental peak overlap. The maximum entropy method (MEM) can be used to extract the maximum amount of information from a limited set of data by maximizing the entropy. The entropy of any trial density is defined relative to a prior density ρ^{prior} as

$$S = -\sum_{i=1}^{N_{\text{pix}}} \rho_i \ln(\rho_i / \rho_i^{\text{prior}}), \qquad (1)$$

where the electron density is sampled at the points of a $N_1 \times N_2 \times N_3 = N_{\text{pix}}$ grid over the unit cell. The prior density or prior can be a constant function, thus called a flat or uniform prior. Alternatively, it can represent any desirable distribution of the available electrons over the unit cell. A particularly useful prior is the procrystal density that is defined as the electron density corresponding to the indepedent atom model (IAM), as it is obtained, for example, from the best Rietveld refined structure model.

The goal of the MEM is to find the electron density that maximizes the entropy S subject to various constraints. The method of undetermined Lagrange multipliers is used to find the maximum of $Q = S - \lambda_N C_N - \lambda_F C_F - \lambda_G C_G$, with the Lagrange multipliers λ and the constraints C as defined below. According to $\partial Q/\partial \lambda = 0$, the maximum is reached when the different constraints C_N , C_F and C_G are all zero. All constraints are functions of the electron density. The first constraint is the normalization of the electron density and it is defined as

$$C_{\rm N} = N_{\rm el} - \frac{V}{N_{\rm pix}} \sum_{i=1}^{N_{\rm pix}} \rho_i = 0,$$
 (2)

where $N_{\rm el}$ is the number of electrons in the unit cell of volume V. $\lambda_{\rm N}$ can be eliminated from the MEM equations and does not appear in the iterative procedure. The prior obeys the same normalization; the density of the flat prior thus is $\rho_{\rm i} = N_{\rm el}/V$. The second constraint $C_{\rm F}$ is based on the phased observed structure factors $F_{\rm obs}$ and is defined as

$$C_{\rm F} = -1 + \frac{1}{N_{\rm F}} \sum_{i=1}^{N_{\rm F}} w_i \left(\frac{|F_{\rm obs}(\mathbf{H}_i) - F_{\rm MEM}(\mathbf{H}_i)|}{\sigma(\mathbf{H}_i)} \right)^2 = 0, \quad (3)$$

where $N_{\rm F}$ is the number of isolated phase observed structure factors $F_{\rm obs}(\mathbf{H}_i)$ with standard uncertainties σ (H_i); $F_{\rm MEM}(\mathbf{H}_i)$ are the phased structure factors obtained by Fourier transform of the trial electron density (ρ_i); w_i is the weight factor (its value is unity if no weights are applied).

Overlapping reflections in powder diffraction are managed in the MEM by applying the so-called G-constraint (Sakata *et al.*, 1990), defined as

$$C_{\rm G} = -1 + \frac{1}{N_{\rm all}} \sum_{i=1}^{N_{\rm F}} w_i \left(\frac{|F_{\rm obs}(\mathbf{H}_i) - F_{\rm MEM}(\mathbf{H}_i)|}{\sigma(\mathbf{H}_i)} \right)^2 + \frac{1}{N_{\rm all}} \sum_{j=N_{\rm F}+1}^{N_{\rm all}} \left(\frac{G_{\rm obs}^j - G_{\rm MEM}^j}{\sigma(G_{\rm obs}^j)} \right)^2 = 0,$$
(4)

where $N_{\rm G}$ is the number of overlap groups and

$$G^{j} = \sum_{k=1}^{N_{\rm G}(j)} \left(\frac{m_{k}}{\sum m_{k}} |F(H_{K})|^{2}\right)^{1/2}$$
(5)

with the multiplicity m_k of the reflection k. $N_G(j)$ is the number of reflections contributing to the *j*th overlap group. The error propagation law is used to calculate the standard uncertainty of the group reflection according to

$$\sigma(G) = 1/G(j) \left[\sum_{i} \left(F_j \left(m_j / \sum m_k \right) \sigma(F_j) \right)^2 \right]^{1/2}.$$
 (6)

Convergence of the MEM can only be achieved if a sufficiently large fraction of the reflections is part of the F constraint (Sakata *et al.*, 1990). In our approach, a single Lagrange multiplier has been used for the combined F and G constraint defined as

$$C_{\rm FG} = C_{\rm G} = -1 + \frac{1}{N_{\rm all}} \sum_{i=1}^{N_{\rm F}} w_i \left(\frac{|F_{\rm obs}(\mathbf{H}_i) - F_{\rm MEM}(\mathbf{H}_i)|}{\sigma(\mathbf{H}_i)} \right) + \frac{1}{N_{\rm all}} \sum_{j=N_{\rm F}+1}^{N_{\rm all}} \left(\frac{G_{\rm obs}^j - G_{\rm MEM}^j}{\sigma(G_{\rm obs}^j)} \right) = 0$$
(7)

where $N_{\rm all} = N_{\rm F} + N_{\rm G}$.

The initial electron density or prior density ρ_i^{prior} can be introduced as a flat-prior, as in the present case for the



Figure 1

Flowchart showing the procedure for extracting the different types of structure factors and phases used to reconstruct the different types of electrondensity maps from powder diffraction data. The procedure of the combination of MEM and CF is marked by a dashed line.

<i>c</i> (Å)	8.255 (24)	7.796 (39)
$V(Å^3)$	299.1 (16)	234.4 (20)
F(000)	236	208
Temperature (K)	683	873
Wavelength λ (Å)	0.9224 (2)	0.9224 (2)

Chemical formula

Space group Ź

 $(\sin \theta / \lambda)_{max}$

 $R_{\rm p}$

 $\dot{R_{w}}$

GOF

No. of unique reflections

No. of group reflections

a (Å)

b (Å)

Table 1 Crystallographic information of α -Rb₂C₂O₄, α -Rb₂CO₃ and δ -K₂C₂O₄ as obtained from Rietveld refinements

 α -Rb₂CO₃

P63/mmc

5.892 (24)

5.892 (24)

7.796 (39)

0.78630

0.0120

0.0192

0.98

39

6

 $\delta - K_2 C_2 O_4$

10.930 (28)

6.1284 (15)

3.4830 (9)

233.3 (17)

0.9224(2)

0.78630

0.0138

0.0183

1.10

164

295

68

9

Pbam

2

 α -Rb₂C₂O₄

P63/mmc

6.468 (18)

6.468 (18)

8.255 (24)

0.78630

0.0105

0.0139

0.79

40

6

description	of	disorder	or	in	the	case	of	anharmonic
temperature	vib	rations wh	en t	he e	exact	positic	on of	f the atoms is
the quantity	of	interest. C)n tl	he c	other	hand,	the	procrystal is
used as the	prio	r density	whe	n th	e qu	antity	of ii	nterest is the
redistributio	n of	electron	den	sity	due 1	to cher	mica	l bonds.

2.2. Charge flipping for powder (pCF)

Charge flipping is an algorithm that was recently proposed for ab initio structure determination from single-crystal and powder X-ray diffraction data (Oszlányi & Sütő, 2004, 2005; Palatinus, 2004; Baerlocher et al., 2007). The method is based on an iterative procedure, whereby in each cycle density values smaller than some positive threshold δ values are given the opposite sign. The result is an electron-density map and a set of phases for the observed structure factors. With powder diffraction data, the Le Bail method can be used to extract the starting values of the structure-factor amplitudes. Histogram matching (Zhang & Main, 1990) can be used to repartition



Figure 2

Two perspective projections each of the disordered oxalate and carbonate dianions of (a) α -Rb₂C₂O₄ and (b) α -Rb₂CO₃, and of the ordered, planar oxalate unit in (c) δ -K₂C₂O₄ dianions.

overlapping reflections through the step of density modification. We have used this version of charge flipping (pCF) for the reconstruction of phases of structure factors (Baerlocher et al., 2007).

2.3. Combination of the MEM and charge flipping

Much attention has been given to the model bias of the MEM. Both the values of the observed structure-factor amplitudes and the phases could be affected by the model, thus leading to biased electron densities. To overcome this problem, pCF is used to solve the phase problem. It is assumed that the structure-factor amplitudes have been extracted (using the Le Bail method or a similar algorithm). The reflection phases obtained by pCF can be combined with either the observed structure factors obtained from Rietveld refinement or the structure factors extracted from a Le Bail fit. In the first case the Rietveld model can be tested for phase biasing effects, while for the second case the resultant MEM map is considered completely unbiased by a structure model, since the phases and structure-factor amplitudes are obtained directly from the data. Uncertainties or model bias on amplitudes are removed by the use of combined F and Gconstraints together with phases from CF. In Fig. 1 the dashed outline refers to the procedure of MEM and CF; the two different combinations are designated by (A) and (B).

3. Experimental

3.1. X-ray powder diffraction

In situ X-ray powder diffraction data of potassium and rubidium oxalates and their corresponding carbonates as decomposition products were collected at high temperature using a wavelength of 0.9224 (2) Å in transmission geometry with a small hot environment cell at beamline X7B at the National Synchrotron Light Source (NSLS) at Brookhaven

> National Laboratory. The samples were contained in sealed quartz capillaries (diameter 0.5 mm). Details of the experimental setup as well as the description of the different crystal structures have been previously described (Dinnebier et al., 2005). For the present study, three phases were selected: the highest-temperature phases α -Rb₂C₂O₄ and α -Rb₂CO₃, and the room-temperature phase δ - $K_2C_2O_4$. The δ phase of potassium oxalate crystallizes in the space group Pbam, with the ordered oxalate anion in planar conformation (Fig. 2c). The oxalate anions in the α phase of rubidium oxalate in the space group P6₃/mmc are staggered and orientationally and conformationally disordered (Fig. 2a). Good Rietveld refinements are achieved assuming a

Table 2

Characteristics of the MEM calculations.

All calculations are based on the Sakata-Sato algorithm and a flat prior (§2.1).

Chemical formula	δ -K ₂ C ₂ O ₄	α -Rb ₂ C ₂ O ₄	α -Rb ₂ CO ₃
Number of pixels	$108 \times 64 \times 36$	$96 \times 96 \times 108$	$54 \times 54 \times 72$
Pixel size (\mathring{A}^3)	$0.101 \times 0.095 \times 0.096$	$0.067 \times 0.067 \times 0.076$	$0.109 \times 0.109 \times 0.108$
Electrons per unit cell	164	236	208
Lagrange multiplier λ	0.05	0.05	Automated
$R_{\rm E}/R_{\rm wF} \left(\rho_{E}^{\rm MEM}\right)$	0.00626/0.00351	0.0181/0.0054	0.0203/0.0051
$R_{\rm F}/R_{\rm wF} \left(\rho_{E}^{\rm MEM}\right)$	0.0066/0.0035	0.0236/0.0054	0.0174/0.0051
$R_{\rm E}/R_{\rm wE} R_{\rm G}/R_{\rm wG} (\rho_{E,\pm,\pm,G}^{\rm MEM})$	0.0073/0.0035, 0.0055/0.0042	0.0210/0.0050, 0.0332/0.0105	0.0180/0.0050, 0.0148/0.0127
$R_{\rm E}/R_{\rm wE} R_{\rm G}/R_{\rm wG} (\rho_{\rm MEM}^{\rm MEM})$	0.0149/0.0117, 0.0208/0.0124	0.0329/0.0062, 0.0260/0.0077	0.0132/0.0052, 0.0277/0.0172
$R_{\rm E}/R_{\rm wE} R_{\rm C}/R_{\rm wC} (\rho_{\rm LeBail+G}^{\rm MEM})$	0.0143/0.0118, 0.0176/0.0117	0.0373/0.0062, 0.0286/0.0085	0.0173/0.0049, 0.0305/ 0.0192
$R_{\rm F}/R_{\rm wE}$, $R_{\rm G}/R_{\rm wG}$ ($\rho_{\rm LeBail+CF+G}^{\rm MEM}$)	0.0057/0.0035, -/-	0.0217/0.0051, 0.0311/0.0097	0.0167/0.0052, 0.0041/0.0036
$R (\text{diff } \rho_{E}^{\text{MEM}})$	0.029162	0.064700	0.05399
$R \left(\text{diff } \rho_{\text{MEM}}^{\text{MEM}} \right)$	0.055154	0.070854	0.06054
$R \left(\text{diff } \rho_E^{\text{MEM}} \right)$	0.1663	0.096320	0.08235
$R \left(\text{diff } \rho_E^{\text{MEM}} \right)$	0.1984	0.108683	0.0949
$\frac{R \left(\text{diff}_{-}\rho_{F_{\text{obs}+\text{CF+G}}}^{\text{MEM}} \right)}{R \left(\text{diff}_{-}\rho_{F_{\text{obs}+\text{CF+G}}}^{\text{MEM}} \right)}$	0.2005	0.082994	0.08857

24-fold disordered oxalate anion which is rotated around an axis perpendicular to the principal axis in addition to having rotated carboxyl groups, thus leading to a quasi-spherical entity. The C-C bond of the anion is diagonally oriented with respect to the *ab* plane leading to 12-fold disorder, which is doubled by a mirror plane perpendicular to the *c* axis. The



Figure 3

(a) Rietveld plots for α -Rb₂C₂O₄, α -Rb₂CO₃ and δ -K₂C₂O₄. Selected regions of (b) Rietveld and (c) Le Bail plots of the powder patterns of (a) showing overlapping reflections and their decomposition.



Figure 4

Electron densities obtained by charge flipping for (a) δ -K₂C₂O₄ (iso-level 2.5 e Å⁻³); (b) α -Rb₂C₂O₄ (iso-level 0.25 e Å⁻³); (c) α -Rb₂CO₃ (iso-level 0.6 e Å⁻³). Colored spheres represent atomic positions of the Rietveld model.



Figure 5

Perspective views and corresponding two-dimensional projections of the procrystal electron densities of ordered δ -K₂C₂O₄ at 295 K (*a*), (*b*), disordered α -Rb₂C₂O₄ at 683 K (*c*), (*d*), and disordered α -Rb₂CO₃ at 873 K (*e*), (*f*). The following parameters have been used: (*a*) iso-level 1.0 Å⁻³, volume/area = 30.1/142.2 (Å³/Å²); (*b*) projection along [001], contours 0.2 e Å⁻³, cut-off at 2.5 e Å⁻³, schematized oxalate dianion; (*c*) iso-level 0.6 Å⁻³, volume/area = 95.56/250.930 (Å³/Å²); (*d*) projection along [110], contours 0.1, cut-off 3 e Å⁻³; (*e*) iso-level 0.6, volume/area = 63.85/195 (Å³/Å²); (*f*) projection along [110], contours 0.1, cut-off 3 e Å⁻³. Colored spheres represent atomic positions of the Rietveld model.

refined orientation of the oxalate dianion deviates from the ideal staggered conformation by approximately 12°. Since the mobility of the cations increases considerably with temperature, anisotropic a.d.p.s must be introduced. The crystal structure of α -Rb₂CO₃ (Fig. 2*b*) is isotypic to that of α -Rb₂CO₄ except for the fact that the carbonate group has no internal degree of freedom leading to pure orientational disorder. Crystallographic data for all three phases are listed in Table 1.

3.2. Rietveld refinements with local site symmetry

Pseudo-rigid-body Rietveld refinements were performed using JANA2000 (Petricek et al., 2000) for the crystal structures of α -Rb₂C₂O₄ and δ -K₂C₂O₄ (Fig. 3). The oxalate molecular anion was defined in a Cartesian coordinate system assuming the point group 222 as the internal symmetry of the anion with the C-C bond aligned along a twofold axis. The torsion angle, isotropic a.d.p.s of O and C atoms as well as anisotropic a.d.p.s for the rubidium cations were subjected to Rietveld refinement.

3.3. Charge flipping

Phases of reflections were reconstructed by the charge-flipping (CF) algorithm (Oszlányi & Sütő, 2004, 2005; Palatinus, 2004) combined with the method of histogram matching for repartitioning of the overlapping reflections (Zhang & Main, 1990; Baerlocher et al., 2007). The structure factors were extracted by the Le Bail method. All calculations of CF were performed starting from random phases using the computer program Superflip (Palatinus & Chapuis, 2007). The method of low-density elimination the (LDE; Shiono & Woolfson, 1992) was used from the beginning of the calculations for the three structures to modify the density in real space. The histogrammatching method was used to repartition the overlapping reflections only in case of the disordered crvstal structures

(Figs. 4b and c). The histogram was created according to the chemical composition of the unit-cell content, taking into account the number of formula units per unit cell. A value of 0.2 for the parameter 'weak-ratio' was found to give the best performance. We also found that the histogram-matching method did not work well in the case of heavily overlapping reflections, and led to divergence. However, the lowdensity elimination method (LDE) already has the ability to reconstruct the phases of the model (Fig. 4a).

Comparison of the reflection phases resulting from CF with the phases corresponding to the structure models obtained by Rietveld refinements for the three structures shows that among 86/51/44 reflections for δ -K₂C₂O₄/ α -Rb₂C₂O₄/ α -



Figure 6

Difference electron densities: $\rho_{\text{calc}}^{\text{MEM}} - \rho^{\text{procrystal}}$ for the crystal structures of ordered δ -K₂C₂O₄ in (*a*) and disordered α -Rb₂C₂O₄, α -Rb₂CO₃ in (*b*) and (*c*). Colored spheres represent atomic positions of the Rietveld model.



S3, S4 and S5¹). However, the resultant electron densities are not sufficiently accurate for the investigation of details of the crystal structures (Fig. 4): the reconstructed electron densities are affected by inaccurate estimates of the structure-factor amplitudes, possible errors of the scale factors and the series effect termination in the Fourier transforms. Despite these drawbacks, CF is a fast method to reconstruct the approximate image of the electron density using only the lattice parameters and the values of structure-factor amplitudes taken from a Le Bail fit.

3.4. MEM calculations

Calculations according to the MEM were performed with the computer program BayMEM (van Smaalen et al., 2003), employing the Sakato-Sato algorithm (Sakata et al., 1990) and a uniform prior electron density ($\rho_i^{\text{prior}} = \text{constant}$). Unit weights were used. Optimal values for the Lagrange multiplier λ_{FG} were obtained by trial and error. Values of λ_{FG} that were very small initially were increased until each step of the iteration involved a decrease of the value of the constraint by 10%. Too small values of λ_{FG} lead to very slow convergence. Too large values of λ_{FG} reveal themselves through divergence of the iterative procedure or at least an increase of the constraint value for several cycles of the iteration. The latter type of densities are unreliable and were not used. We found that a starting value of $\lambda = 0.05$ yielded the best convergence for all three data sets; an automated adjustment of λ during iterations was

Figure 7

MEM electron-density maps for the crystal structures of ordered δ -K₂C₂O₄ in (*a*) and disordered α -Rb₂C₂O₄, α -Rb₂CO₃ in (*b*) and (*c*) with different data subsets of F_{calc} (*a*1, *b*1, *c*1), F_{obs} (*a*2, *b*2, *c*2), F_{obs} + G constraints (*a*3, *b*3, *c*3) and F_{LeBail} + G constraints (*a*4, *b*4, *c*4). The iso-levels and the values of volume and area of the densities surfaces at the special iso-surfaces are displayed.

 Rb_2CO_3 only 6/4/4 mismatches between the signs of the reflection phases exist for these centrosymmetric structures. All of these reflections are weak (supplementary data, Tables

¹ Supplementary data for this paper are available from the IUCr electronic archives (Reference: KD5036). Services for accessing these data are described at the back of the journal.

Table 3

Observed $(G_{obs}|F_{obs} \text{ and } G_{LeBail}|F_{LeBail})$ and MEM-reconstructed $(G_{MEM}|F_{MEM})$ structure factors for reflections in four selected *G* groups.

Results of two MEM calculations on α -Rb₂CO₃ are given: F_{obs} with φ^{CF} (A in Fig. 1) and F_{LeBail} with φ^{CF} (B in Fig. 1).

	$F_{\rm obs} + G {\rm corr}$	istraint	$F_{\text{LeBail}} + G$ constraint		
hkl	$G_{\rm obs} F_{\rm obs}$	$G_{\rm MEM} F_{\rm MEM} $	$G_{\rm LeBail} F_{\rm LeBail} $	$G_{\rm MEM} F_{\rm MEM} $	
	11.54508	11.37630	11.25819	11.063571	
205	1.42180	1.39721	11.41853	0.389331	
412	14.10400	13.89800	11.17715	13.54725	
	8.14108	8.576332	7.655055	7.851617	
106	7.00589	6.948771	7.869980	5.940244	
401	-9.13630	-9.940902	-7.433918	-9.38132	
	4.84396	4.450077	4.675526	5.212715	
315	0.00000	0.233524	4.711191	1.771974	
402	8.39000	7.700681	4.603366	8.673950	
	5.34194	4.983846	5.000933	4.750375	
206	5.12069	4.988064	4.949198	4.828800	
521	-5.44920	-4.981735	-5.026601	-4.71067	

allowed for α -Rb₂CO₃ (Table 2). Observed structure-factor amplitudes were obtained up to the experimental limit of $\sin(\theta)/\lambda = 0.78630 \text{ Å}^{-1}$ by model-biased decompositions from the final Rietveld refinements. Phased observed structure factors F_{obs} were then derived according to the procedure of Bagautdinov *et al.* (1998), and were used in the *F* constraints. The group amplitudes of groups of overlapping reflections (*G* amplitude) were calculated from the structure factors according to (5) with standard uncertainties (σ group) according to (6). These values were used in the *G* constraints. Alternatively, structure-factor amplitudes extracted from Le Bail fits were scaled according to the absolute scale of the Rietveld refinement. The structure factors obtained from Le Bail fit F_{LeBail} were corrected for the anomalous scattering effect according to the same procedure which is used for the correction of F_{obs} , *i.e.* applying the ratio of $F_{calc}(hkl)$, without AS)/ $F_{calc}(hkl)$, with AS), where AS is the anomalous scattering effect. Very small changes in the values of F_{leBail} were thus found, which is reasonable because of the small anomalous scattering contributions (Rb has f'' = 0.66 and K has f'' = 0.54). The effects of possibly erroneous scale factors were tested by additional calculations employing scale factors larger and smaller by 5%. Despite minor differences in the densities, the main features of the disorder remained unaffected. Electron densities from *BayMEM* were reformatted in the unformatted ASCII file type of *X-PLOR*, and three-dimensional visualizations were produced by the *UCSF* software Chimera (Pettersen *et al.*, 2004). A small *Fortran* program was written for the calculation of the difference densities in three-

4. Electron densities

dimensions.

4.1. Procrystal density

Electron densities corresponding to the refined structure models (the procrystal densities $\rho^{\text{procrystal}}$; see §2.1) were calculated for the three compounds by the computer program PRIOR (van Smaalen et al., 2003). The densities were sampled on a suitably fine grid (Table 2). Atomic positions and a.d.p.s were used from the final rigid-body Rietveld refinements (§3.2 and Table 1). Selected sections of the densities were visualized by contour plots generated with JANA2000, and threedimensional visualizations were produced with the computer program Chimera (Pettersen et al., 2004; Fig. 5). It is worth noting that this type of density map provides an approximation to the real electron density, while deformations of the electron density due to chemical-bonding effects or disorder beyond the model are not considered. The procrystal density has been used as a reference model for MEM calculations and to visualize structural differences between MEM densities and the procrystal structure.



Figure 8

MEM electron densities of ordered δ -K₂C₂O₄ in (*a*) and disordered α -Rb₂C₂O₄, α -Rb₂CO₃ in (*b*) and (*c*) based on reflection phases from charge flipping with two different types of amplitudes. (*a*1), (*b*1), (*c*1) |*F*_{LeBail}|, and (*a*2), (*b*2), (*c*2) |*F*_{obs}|.

The MEM densities ρ_{calc}^{MEM} have been calculated with F constraints based on the calculated structure factors F_{calc} of the structure models obtained by Rietveld refinement (Fig. 1). This type of map should reproduce the structure model, but differences will occur because of series termination effects, i.e. structure factors have been used up to the experimental limit of $\sin(\theta)/\lambda = 0.7863 \text{ Å}^{-1}$ only. Furthermore, MEM densities may deviate from the true densities due to inadequacies of the MEM itself, in particular due to inadequacies related to the standard uncertainties of the structure factors and the choice of weights in the *F* constraints [see (3)]. The differences between reconstructed and true densities can be analyzed by consideration of the difference densities $\rho_{\rm calc}^{\rm MEM} - \rho^{\rm procrystal}$, as shown in Fig. 6.

Although the difference densities are small (real-space agreement factors between $\rho_{\rm calc}^{\rm MEM}$ and $\rho^{\rm pro}$ are R = 0.0598 for α -Rb₂C₂O₄ and R = 0.0416 for α -Rb₂CO₃), they can be of the same order of magnitude as the effects being studied. Under





Difference MEM electron densities: $(\rho_{(1)}^{\text{MEM}} - \rho_{\text{calc}}^{\text{MEM}})$ of ordered δ -K₂C₂O₄ in (*ai*) and disordered α -Rb₂C₂O₄, α -Rb₂CO₃ in (*bi*) and (*ci*), where *i* = 1 refers to F_{obs} (first row), *i* = 2 refers to F_{obs} with *G* constraints (second row) and *i* = 3 refers to F_{LeBail} with *G* constraints (last row). Colored spheres represent atomic positions of the Rietveld model.



Figure 10

Two-dimensional sections of MEM and difference MEM maps based on F_{obs} -G constraints for the three structures. (*a*1) and (*a*2) are sections along and perpendicular (includes only C and O atoms) to the mirror plane containing the planar oxalate dianion; contours are drawn at 0.5 e Å⁻³ with a cut-off of 5 e Å⁻³. (*a*3) is a section along the 002 plane of the difference MEM density; contours are at 0.15 e Å⁻³. (*b*1, *b*2) and (*c*1, *c*2) are sections parallel to the 110 plane for the MEM map (contour lines at 0.1 and cut-off 3 e Å⁻³) and difference MEM maps [+ ve (solid lines), -ve (dashed lines) contours are at 0.15 e Å⁻³] for rubidium oxalate and rubidium carbonate.

the assumption that series-termination effects and other errors will have similar values in different maps, we have employed $\rho_{\text{calc}}^{\text{MEM}}$ as a reference density for analyzing the differences between the various experimental MEM densities and the density of the IAM model. This assumption is justified by the fact that all MEM maps have been obtained with data up to the same resolution and with the same standard uncertainties and the same weighting scheme.

The MEM densities have been calculated with F constraints based on the phased observed structure factors F_{obs} [Figs. 1, 7(a2), (b2) and (c2)]. Especially in the case of powder diffraction data this kind of map is biased by the model, because calculated structure factors of the model are used to decompose overlapping reflections into contributions of individual reflections. The number of overlapping reflections is small for the two disordered structures. but it is larger for the lowersymmetry structure δ -K₂C₂O₄. Therefore, additional MEM densities have been calculated with F constraints based on the phased observed structure factors $F_{\rm obs}$ of well resolved reflections and G constraints based on the group amplitudes of overlapping reflections (Fig. 1). The densities $\rho_{\rm obs+G}^{\rm MEM}$ are given in Figs. 7(a3), (b3) and (c3). The capability of the MEM to reconstruct the values of structure-factor amplitudes and their phases of individual reflections from the Gconstraint is illustrated in Table 3 for four overlap groups in the diffraction data of α -Rb₂CO₃.

Unbiased observed structurefactor amplitudes are those extracted from Le Bail fits to the diffraction data, *i.e.* by a Le Bail decomposition disregarding any structural information. This procedure provides amplitudes as they would have been directly measured by single-crystal diffraction, except that the Le Bail procedure can only provide accurate amplitudes for resolved reflections. For each group of overlapping reflections an accurate value is obtained for the sum of intensities. Therefore, useful MEM calculations can only be performed with a combination of *F* and *G* constraints, resulting in the densities $\rho_{\text{LeBail+G}}^{\text{MEM}}$ [Figs. 1, 7(*a*4), (*b*4) and (*c*4)]. The fourth and fifth columns in Table 3 clearly show that the MEM has successfully partitioned the groups of overlapping reflections into contributions of individual reflections and that the MEM leads to the same phases as have been obtained by charge flipping. For all four overlap groups the



Figure 11

Difference MEM electron densities: $(\rho_{(1)}^{\text{MEM}})$ of ordered δ -K₂C₂O₄ in (*ai*) and disordered α -Rb₂C₂O₄, α -Rb₂CO₃ in (*bi*) and (*ci*), where *i* = 1 refers to F_{LeBail} (first row) and *i* = 2 refers to F_{obs} (second row). In both cases the phases obtained from CF and applying *G* constraints for the three compounds. Colored spheres represent atomic positions of the Rietveld model.



Figure 12

Two-dimensional sections of the MEM and difference-MEM maps based on F_{LeBail} -G constraints and phases from CF for the three compounds. (a1) and (a2) show sections along and perpendicular (includes only C and O atoms) to the mirror plane containing the planar oxalate anion. Contours are at 0.5 with a cut-off of 5 e Å⁻³. (a3) is a section along the 002 plane for the difference MEM, contour lines at 0.15 e Å⁻³. (b1, b2) and (c1, c2) are the sections along the 110 plane for the MEM map (contour lines at 0.1 e Å⁻³ with a cut-off 3 e Å⁻³) and difference MEM maps [+ ve (solid lines), -ve (dashed lines) contours are drawn at 0.15 e Å⁻³] for rubidium oxalate and rubidium carbonate.

two reflections have almost perfect overlap, the Le Bail decomposition has thus given nearly equal magnitudes for the structure factors of the two reflections in each group (Table 3).

4.3. MEM densities based on charge flipping

The phased observed structure factors obtained after Rietveld refinement are biased towards the structure model for their amplitudes, while they have phases defined by the model. A model-independent estimate of the reflection phases is obtained by charge flipping, resulting in the set of phases $\varphi_{\rm CF}$ MEM densities have been calculated for the combinations of observed structure-factor amplitudes $|F_{\rm obs}|$ with $\varphi_{\rm CF}$ and of

Le Bail-extracted amplitudes $|F_{\text{LeBail}}|$ with φ_{CF} (Fig. 1). Combinations of *F* and *G* constraints have been used, except in the case of potassium oxalate, which is based exclusively on the *F* constraint (Table 2). The resulting densities are given in Fig. 8.

5. Discussion

Six types of MEM electrondensity maps have been calculated for three compounds with related crystal structures. The MEM densities vary in the amount of bias towards the structure models, which affects the amplitudes and phases of the structure factors. The completely biased densities $\rho_{\text{calc}}^{\text{MEM}}$ (Fig. 1) give an indication of the amount of information that can be extracted by the MEM from the data. Comparison of $\rho_{\text{calc}}^{\text{MEM}}$ [Figs. 7(a1), (b1) and (c1)] with the corresponding model densities [Figs. 5(a)-(f)] shows that the latter are much more structured than the former, especially in the cases of the oxalates. This strongly suggests that the highly structured densities of the models are actually artifacts. A smooth character is also found for the MEM densities based on the experimental data (Figs. 7 and 8). Differences with ρ_{calc}^{MEM} can be analyzed on the basis of difference densities

$$\Delta \rho_{\rm (I)}^{\rm MEM} = \rho_{\rm (I)}^{\rm MEM} - \rho_{\rm calc}^{\rm MEM},\tag{8}$$

where (I) stands for one of the six types of maps: obs, obs + G, Le Bail + G, obs + CF, obs + CF + G and Le Bail + CF + G (Fig. 1). The differences are visualized by plots of iso-surfaces of $\Delta \rho_{(1)}^{\text{MEM}}$ (Figs. 9–12) and they are quantified by the area and volumes of these iso-surfaces as well as by the real-space Rvalue of matching $\rho_{(1)}^{\text{MEM}}$ and $\rho_{\text{calc}}^{\text{MEM}}$ (Table 2). The values for the volume and the area of the iso-surfaces are calculated in a way that the iso-surface value of the difference-MEM map is changed until the density level reaches the densities of the MEM map. This procedure of choosing the iso-surface of difference-MEM maps is used because of the huge differences between the ranges of the difference-MEM maps. This holds especially true when the data originate from Le Bail fits of different uncertainty values, compared with Rietveld refinement. Fig. 7 presents comparative considerations of the electron densities of the three compounds with different data subsets as calculated above; the enclosed volume and area of the iso-surfaces are indicated in the maps. For each compound all four maps have similar appearances, but significant differences are also found between any MEM density and the corresponding reference MEM density based on F_{calc} . These differences can be visualized by difference maps (Fig. 9). Analysis of $\Delta \rho_{\text{obs}}^{\text{MEM}}$, $\Delta \rho_{\text{obs}+G}^{\text{MEM}}$ and $\Delta \rho_{\text{LeBail}+G}^{\text{MEM}}$ leads to the following observations:

(i) In the case of the ordered crystal structure of δ -K₂C₂O₄, non-zero values of difference densities $\Delta \rho_{(1)}^{\text{MEM}}$ are restricted to small regions in space. Although the differences can be relatively high, this indicates that the MEM confirms the structure model in most aspects. Observed features in $\Delta \rho_{(1)}^{\text{MEM}}$ can be artifacts but will also include features due to an inadequate description of thermal motion in the model. The latter interpretation is supported by the fact that differences increase on going from F_{obs} to $F_{\text{obs}} + G$ to $F_{\text{LeBail}} + G$ in the MEM calculation. Large displacements are found for the O atoms in directions out of the mirror plane (Fig. 10*a*2).

(ii) For the disordered, high-temperature phase of α -Rb₂C₂O₄, the range of values of $\Delta \rho_{(1)}^{\text{MEM}}$ is similar as for ordered δ -K₂C₂O₄, but the regions of appreciable difference are much larger than in the ordered case. This finding suggests that the model has not captured the disorder with the same precision that is possible for an ordered structure. In addition to the orientational and conformational disorder of the oxalate anions, $\Delta \rho_{(1)}^{\text{MEM}}$ clearly shows the mobility of the cations. The displacements of Rb2 along the *c* axis are larger than those of Rb1, as is most clearly seen in the (110) sections of $\rho_{(1)}^{\text{MEM}}$ and $\Delta \rho_{(1)}^{\text{MEM}}$ [Figs. 10(*b*1) and (*b*2)].

(iii) For the disordered, high-temperature phase of α -Rb₂CO₃, the same conclusions are obtained as for disordered α -Rb₂C₂O₄, but with larger displacements for Rb2 along the *c* axis compared with Rb2 in α -Rb₂C₂O₄ [Figs. 10(*c*1) and (*c*2)].

A new combination of MEM and CF is presented here. Phases from CF are combined with either the observed structure factors from Rietveld refinements or the observed structure factors obtained by the Le Bail procedure (Fig. 1). The first conclusion is that for all three compounds the main features of the densities are also reproduced by these two maps (Figs. 7 and 8). Turned around, this implies that the completely model-free approach of charge-flipping with Le Bail fitting provides a reasonable-to-good description of the crystal structure, including features due to disorder. Nevertheless, $\Delta \rho_{(I)}^{\rm MEM}$ show significant differences between model and MEM densities:

(iv) For ordered δ -K₂C₂O₄, MEM densities are deformed [Figs. 8(*a*1) and (*a*2)] compared with the previous MEM calculations, which is reflected in the high values of the agreement factor (R = 0.2005 and 0.1984) in both cases of combinations.

(v) For the disordered crystal structures of α -Rb₂C₂O₄ and α -Rb₂CO₃, the agreement factors are slightly higher for the MEM maps obtained by the combinations Le Bail + *G*, obs + CF + *G*, and Le Bail + CF + *G* compared with the maps derived from obs and obs + *G* (Table 2). It therefore seems that the use of Le Bail amplitudes and CF phases is particularly successful in the case of partially ordered structures, *i.e.* in the case of densities that are relatively smooth compared with the peaked densities of well located atoms.

6. Conclusions

The electron densities of ordered δ -K₂C₂O₄, and disordered α - $Rb_2C_2O_4$ and α - Rb_2CO_3 have been successfully reconstructed from X-ray powder diffraction data by the maximum entropy method (MEM). The MEM has been applied in a series of calculations ranging from calculations completely biased by the model to model-free calculations. The first observation is that for each compound all MEM calculations lead to similar densities; the MEM calculations give good representations of the crystal structure including disorder and anharmonic atomic displacements (Figs. 7 and 8). In particular, this implies that crystal structures including positional and orientational disorder of functional groups can be determined from X-ray diffraction data without the need for a structure model: structure-factor amplitudes are obtained by Le Bail fits to the diffraction data and structure-factor phases are determined by charge flipping with histogram matching.

The MEM employs phased structure factors as input (Fconstraint), while part of the diffraction information can be available as sums of intensities of groups of overlapping reflections (G constraint) as is typically available from powder diffraction. The MEM with F_{obs} as 'experimental data' (structure-factor phases from the model and structure-factor amplitudes by Le Bail decomposition biased by the model) leads to densities that differ by the least amount from the model. Stepwise replacement of more of the models by experimental-based information leads to increasing differences to the model densities (Figs. 9-12). Part of these differences will be due to inaccurate values for amplitudes or phases of reflections or to intrinsic features of the MEM related to the use of the G constraint. However, another part these differences will reflect anisotropic atomic displacement parameters (ordered δ -K₂C₂O₄) not used in the model, and

they indicate better representations of the disorder in α -Rb₂C₂O₄ and α -Rb₂CO₃ by the MEM than is given by the models.

The most important result is that completely *ab initio* electron-density distributions have been obtained by the MEM applied to the combination of structure-factor amplitudes from Le Bail fits with phases from charge flipping. This new combination of the MEM and the method of charge flipping can thus be used for the determination of partially ordered crystal structures from powder diffraction data.

Financial support by the Fonds der chemischen Industrie (FCI), the Bundesministerium Bildung und Forschung (BMBF) and the German Science Foundation (DFG) are gratefully acknowledged.

References

- Baerlocher, Ch., McCusker, L. B. & Palatinus, L. (2007). Z. Kristallogr. 222, 47–53.
- Bagautdinov, B., Luedecke, J., Schneider, M. & van Smaalen, S. (1998). Acta Cryst. B54, 626–634.
- David, W. I. F., Shankland, K., McCusker, L. B. & Baerlocher, C. (2002). Editors. *Structure Determination from Powder Diffraction Data*. Oxford University Press.
- Dinnebier, R. E., Schneider, M., van Smaalen, S., Olbrich, F. & Behrens, U. (1999). Acta Cryst. B55, 35–44.
- Dinnebier, R. E., Vensky, S., Jansen, M. & Hanson, J. C. (2005). Chem. Eur. J. 11, 1119–1129.

- Oszlányi, G. & Sütő, A. (2004). Acta Cryst. A60, 134-141.
- Oszlányi, G. & Sütő, A. (2005). Acta Cryst. A61, 147-152.
- Oszlányi, G. & Sütő, A. (2007). Acta Cryst. A63, 156-163.
- Palatinus, L. (2004). Acta Cryst. A60, 604-610.
- Palatinus, L. & Chapuis, G. (2007). J. Appl. Cryst. 40, 786-790.
- Palatinus, L. & van Smaalen, S. (2002). Acta Cryst. A58, 559– 567.
- Petricek, V., Dusek, M. & Palatinus, L. (2000). *JANA*2000. Institute of Physics, Praha, Czech Republic.
- Pettersen, E. F., Goddard, T. D., Huang, C. C., Couch, G. S., Greenblatt, D. M., Meng, E. C. & Ferrin, T. E. (2004). J. Comput. Chem. 25, 1605–1612.
- Sakata, M., Mori, R., Kumazawza, S., Takata, M. & Toraya, H. (1990). J. Appl. Cryst. 23, 526–534.
- Shiono, M. & Woolfson, M. M. (1992). Acta Cryst. A48, 451-456.
- Takata, M. (2008). Acta Cryst. A64, 232-245.
- Takata, M. & Sakata, M. (1996). Acta Cryst. A52, 287-290.
- Takata, M., Umeda, B., Nishibori, E., Sakata, M., Saito, Y., Ohno, M. & Shinohara, H. (1995). *Nature (London)*, **377**, 46–49.
- van Smaalen, S. (2007). *Incommensurate Crystallography*. Oxford University Press.
- van Smaalen, S., Palatinus, L. & Schneider, M. (2003). Acta Cryst. A59, 459–469.
- Vries, R. Y. de, Briels, W. J. & Feil, D. (1994). Acta Cryst. A50, 383– 391.
- Vries, R. Y. de, Briels, W. J. & Feil, D. (1996). Phys. Rev. Lett. 77, 1719–1722.
- Yamamoto, K., Takahashi, Y., Ohshima, K., Okamura, F. P. & Yukino, K. (1996). *Acta Cryst.* A**52**, 606–613.
- Yashima, M. & Tsunekawa, S. (2006). Acta Cryst. B62, 161-164.
- Zhang, K. Y. J. & Main, P. (1990). Acta Cryst. A46, 41-46.